

# REPORT DOCUMENTATION PAGE

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15 March 2000

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Larson, C. William; Harper, Jessica; Presilla-Marquez, J.D. (Schafer Corp.), "Matrix Isolation of Boron and Carbon Vapor. Control of Cluster Formation During Preparation and Annealing"

**10<sup>th</sup> Internat'l IUPAC Conference on High Temperature Materials Chemistry** (Statement A)  
(Juelich, Germany, 10-14 April 2000) (Deadline: 31 March 2000)  
and Seminars at Max Planck Inst., Univ of Dortmund, Univ. of Basel, Apr 4, 7, 18

**Matrix Isolation of Boron and Carbon Vapor.  
Control of Cluster Formation During Preparation and Annealing.**

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Pat Carrick (Chief), Jeff Sheehy (Group Leader), Greg Drake, Hi Young Yoo, Jeffrey Mills, Jerry Boatz, Jessica Harper,  
Karl Christe, Mario Fajardo, Michael Tinnirello, Michelle DeRose, Paul Jones,  
Txomin Presilla (Schafer Corporation) Peter Langhoff, Simon Tam, Suresh Suri, William Wilson,

**10<sup>th</sup> International IUPAC Conference on  
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**Universität Dortmund  
Fachbereich Chemie  
Anorganische Chemie  
Dortmund, Germany  
18 April 2000**

**Kinetics of formation of cyclic  $C_6$  and cyclic  $C_8$  and of  
 $B_JC_{n-J}$  clusters ( $J = 0, 1, 2$ ;  $n = 3-11$ ) in solid argon**

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**Institute für Physikalische Chemie**

**Universität Basel**

**Basel, Switzerland**

**7 April 2000**

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**High Energy Density Matter (HEDM) Research Group**

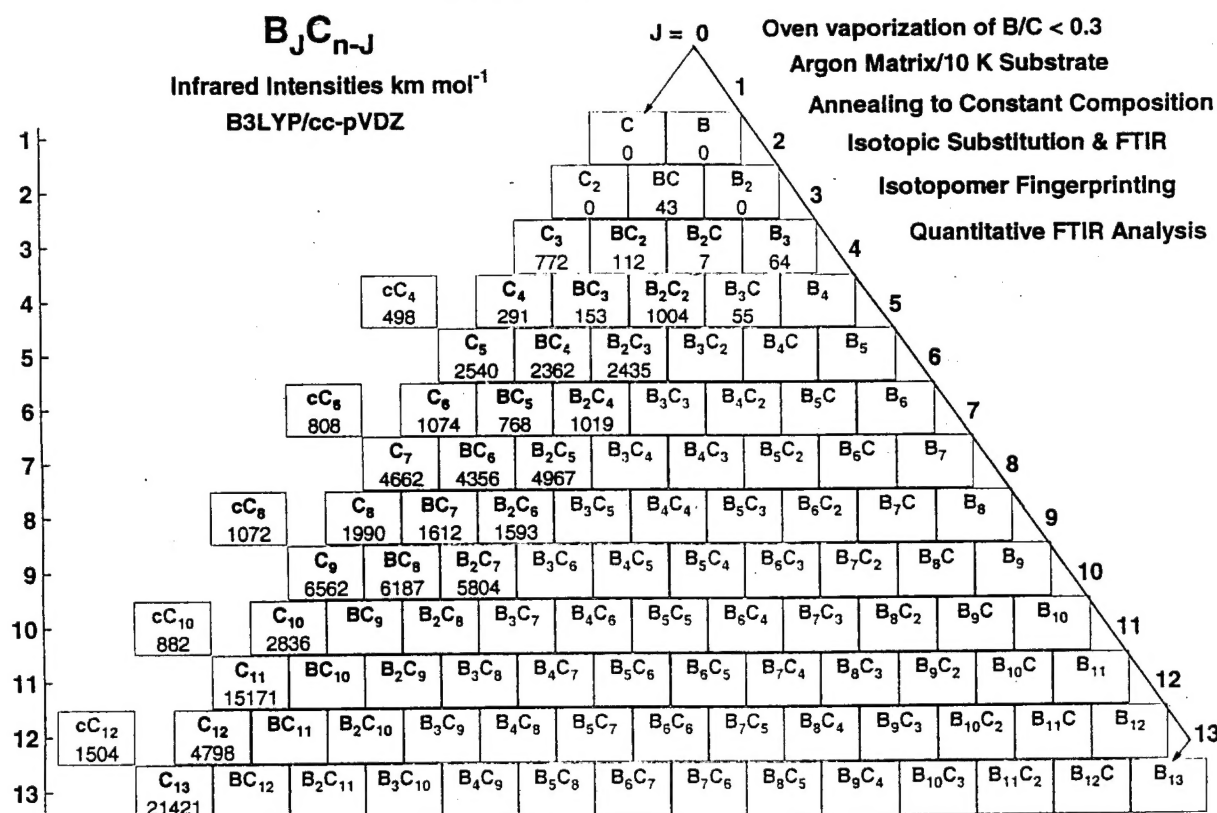
Pat Carrick (Chief), Jeff Sheehy (Group Leader), Greg Drake, Hi Young Yoo, Jeffrey Mills, Jerry Boatz, Jessica Harper,  
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Txomin Presilla (Schafer Corporation) Peter Langhoff, Simon Tam, Suresh Suri, William Wilson,

**Max Planck Institute für Kernphysik**

**Heidelberg, Germany**

**5 April 2000**

## GOAL - 5% atoms in matrix



## Goal

Production of cryogenic HEDM with five mole percent atoms.

## Objective

Characterization of source and quantitative analysis of  $B_J C_{n-J}$

## Approach

Production of HEDM by evaporation of boron with high-temperature graphite furnace and co-deposition of vapor with argon on a cold (10 K) surface

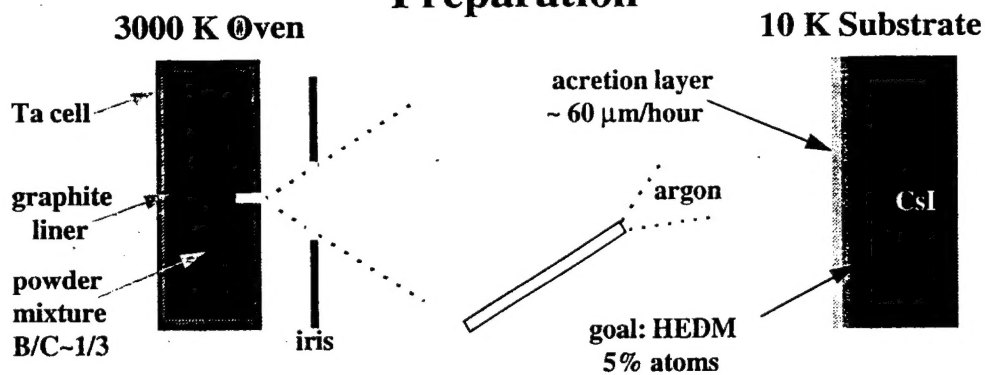
Identification and quantitative analysis of  $B_J C_{n-J}$  species ( $n \geq 3$ ,  $J = 0$  to 2) by FTIR spectroscopy and *ab-initio* calculations.

Quantitative measurement of distributions of  $B_J C_{n-J}$  species produced upon deposition and after annealing to a constant composition.

Determine absolute column densities (molecules  $\text{cm}^{-2}$ ) from Beer's law:

$$\langle \rho_i \rangle = 2.303 A_{\text{exp}} / I_{\text{theory}}$$

## Preparation

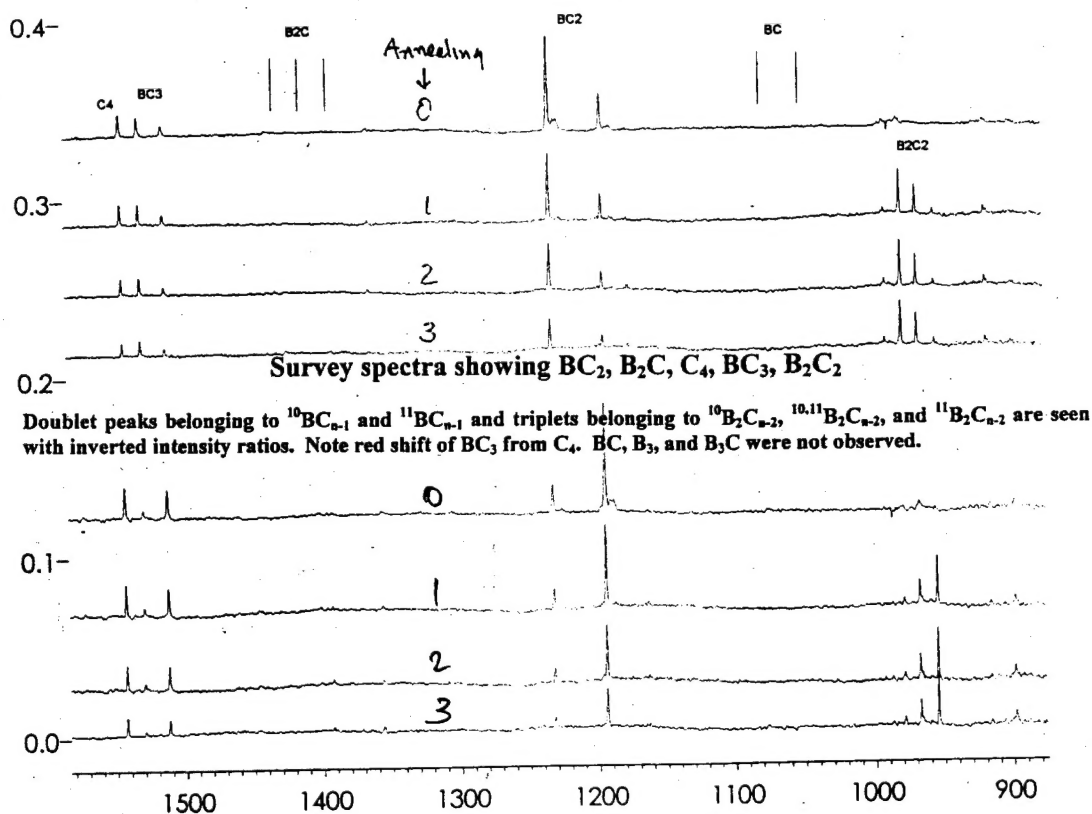


## Annealing

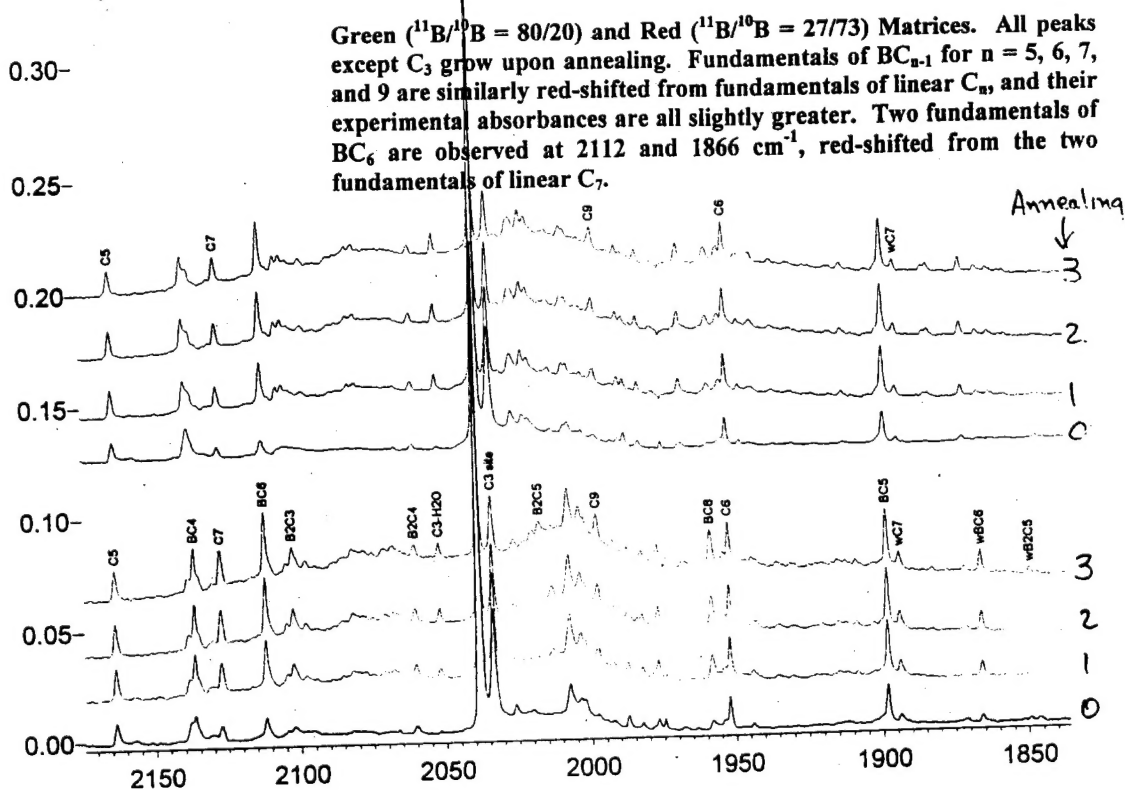
<u>a0</u> 10 K	<u>a3</u> 32.5 K, 60 s	<u>a6</u> 40.0 K, 20 s
<u>a1</u> 27.5 K, 120 s	<u>a4</u> 35.0 K, 45 s	sublimation
<u>a2</u> 30.0 K, 90 s	<u>a5</u> 37.5 K, 20 s	rate ~ 1 $\mu\text{m}/\text{s}$

## Precision matched pair of matrices

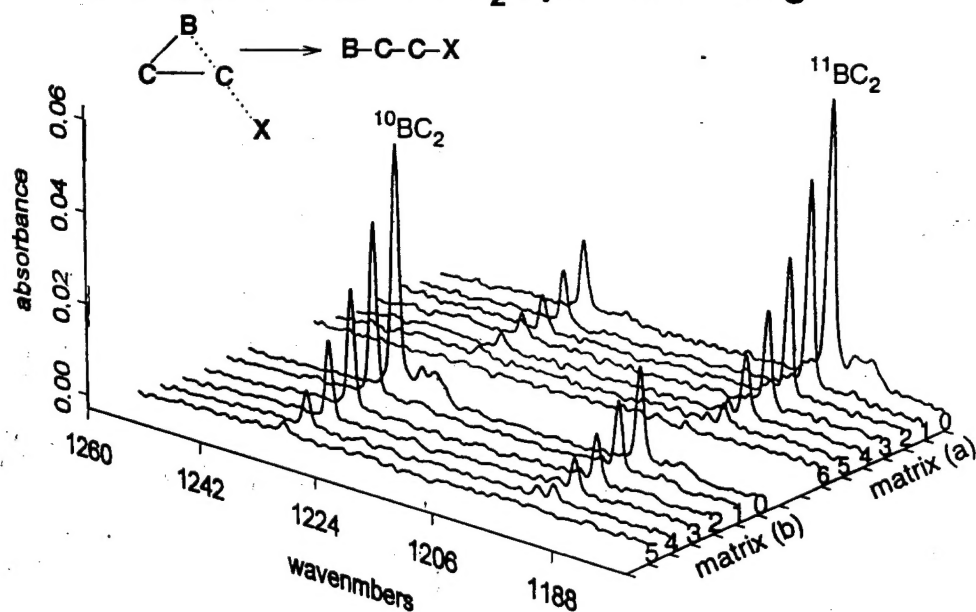
Green Matrix	$^{11}\text{B}/^{10}\text{B} = 80/20$	enhanced $^{11}\text{B}_j\text{C}_{n-j}$
Red Matrix	$^{11}\text{B}/^{10}\text{B} = 27/73$	enhanced $^{10}\text{B}_j\text{C}_{n-j}$



### Survey spectra of precision matched matrices showing larger clusters $B_JC_{n-J}$ , $n > 4$ , $J = 0, 1, 2$ in original matrices and after three annealings.

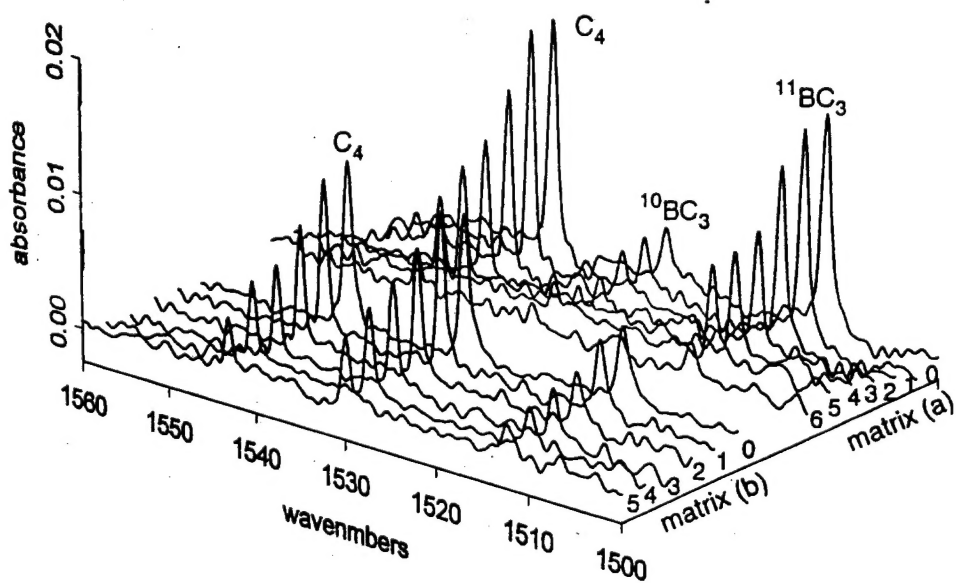
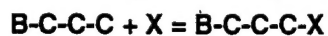


## Disappearance of $BC_2$ upon annealing



BC2rg3D Mar. 8, 2000 9:10:18 AM

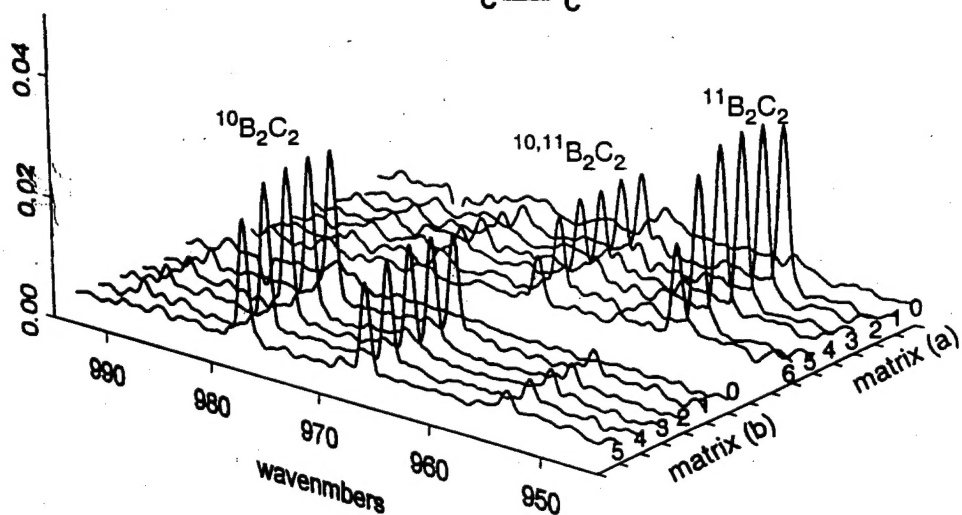
## Disappearance of $BC_3$ and $C_4$ upon annealing



BC3rg3D2.aug Mar. 8, 2000 9:34:49 AM

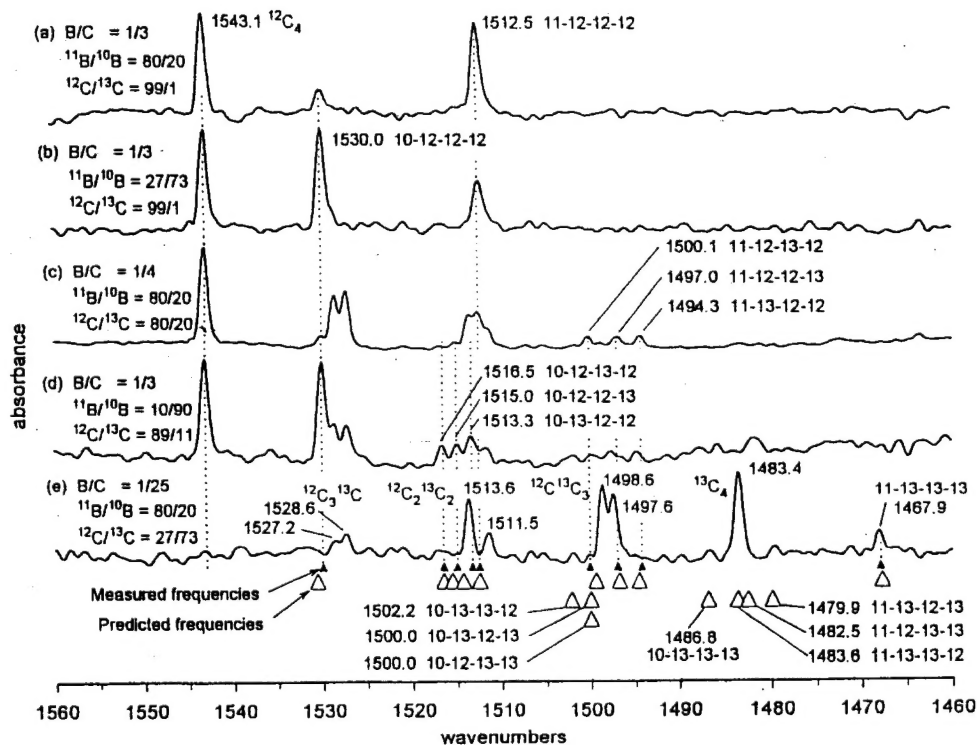


## Growth of $B_2C_2$ upon annealing

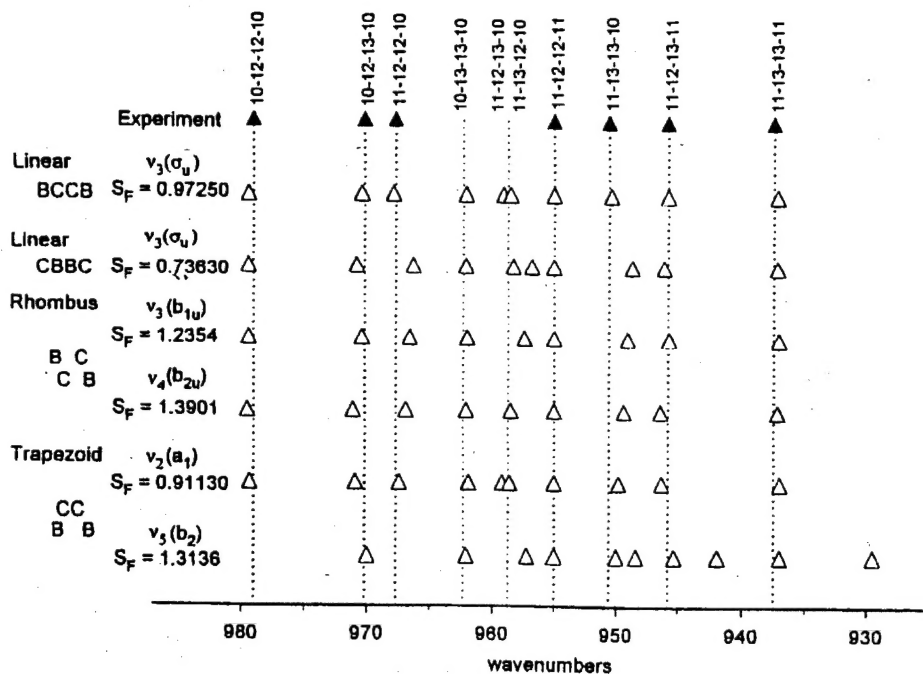


B2C2-3D2 Mar. 8, 2000 7:51:45 AM

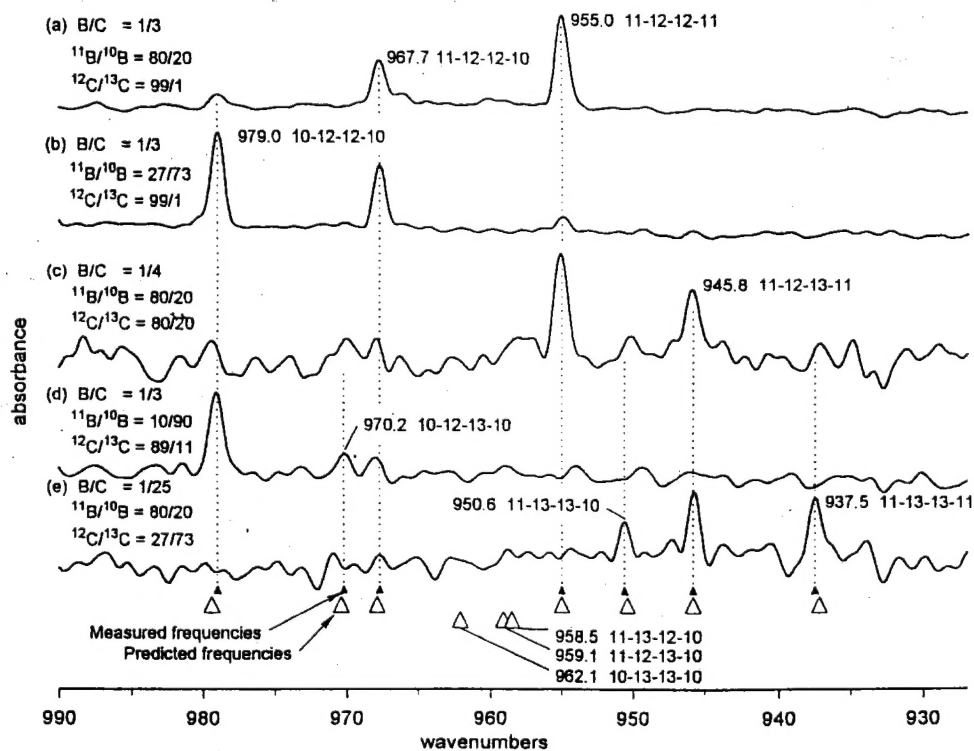
## Identification of 9 of the 16 isotopomers of linear BCCC in 5 matrices.



Four minimum energy geometries of  $B_2C_2$  produce similar isotopomer fingerprints.  
 Scale factor ( $S_F$  = measured frequency/theoretical frequency) of linear BCCB = 0.97250.



Identification of 7 isotopomers of the 10 isotopomers of BCCB in 5 matrices.



# $B_J C_{n-J}$ Annealing Study

## Peak by Peak

### Detail

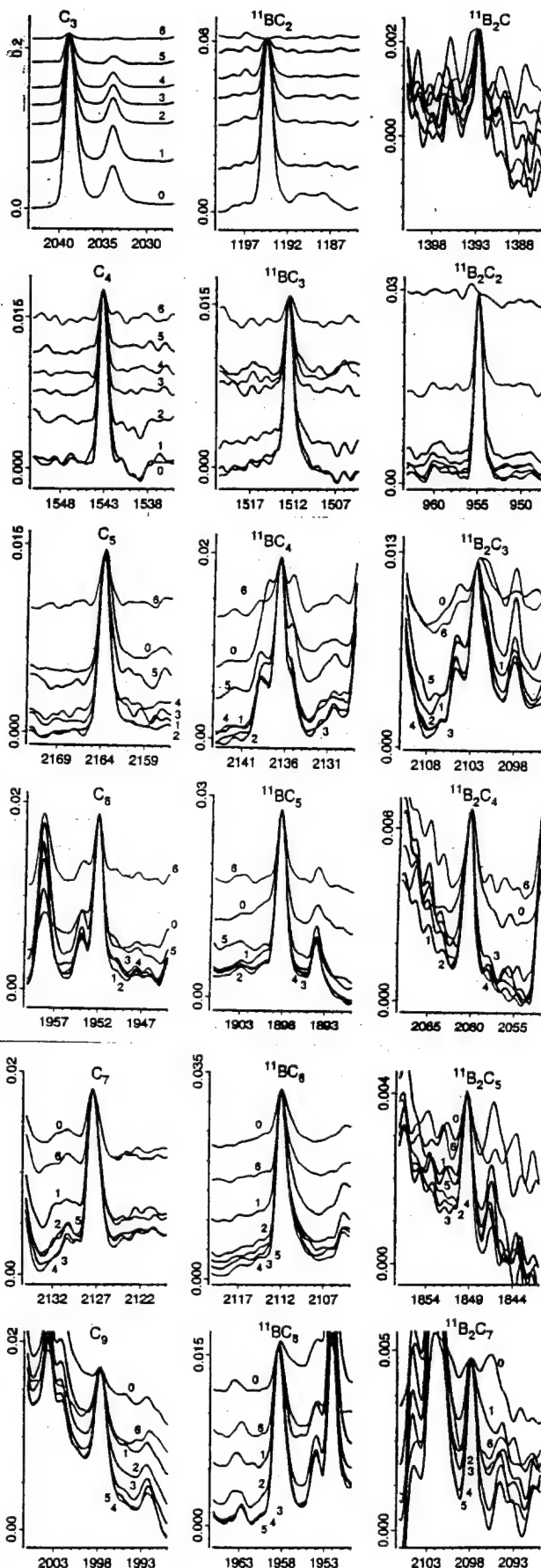
In first annealing  $C_4$  and  $BC_3$  remain constant and  $B_2C_2$  grows from undetectable to  $S/N \sim 100$ .  $B_2C_2$  seems inert.

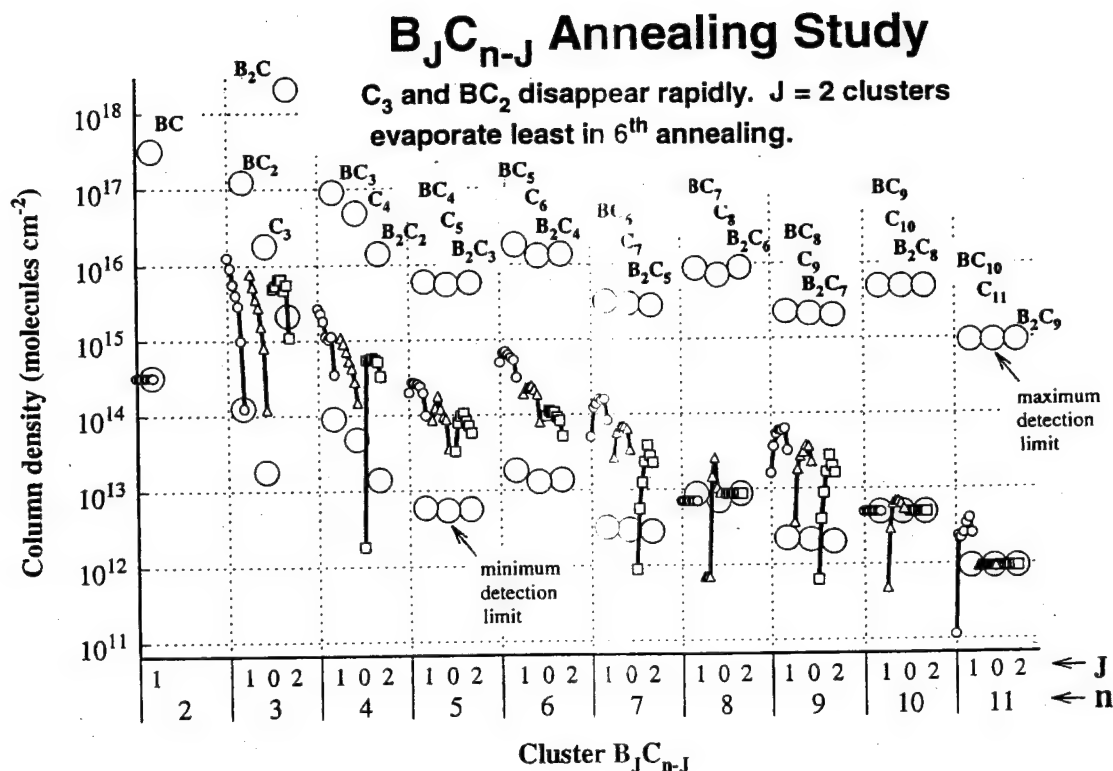
All pentamers grow in first annealing.  $B_2C_3$  grows most;  $C_5$  and  $BC_4$  evaporate most in 5<sup>th</sup> annealing.

Hexamers with boron grow most; evaporation of  $C_6$  in 6<sup>th</sup> annealing is greatest.

In 5<sup>th</sup> annealing,  $J = 0$  evaporates more than  $J=1$ , and  $J = 1$  evaporates more than  $J = 2$ . A boron rich layer appears to be capped by a carbon rich layer.

In 6<sup>th</sup> annealing, where almost half the matrix sublimed, the  $J = 2$  clusters evaporate the least. A boron rich layer appears to be capped by a carbon rich layer.





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## Results and Discussion

Linear C<sub>3</sub>, cyclic BC<sub>2</sub>, and cyclic B<sub>2</sub>C, constituted about 80% of the total observable boron and carbon in the initially deposited matrix, but B<sub>3</sub> was not observed. If B<sub>3</sub> were present, its concentration fell below the detection limit of the system. The measured trimer distribution in the initially formed matrices was  $\rho(C_3) : \rho(BC_2) : \rho(B_2C) : \rho(B_3) \sim 1 : 1.5 : 0.5 : < 0.05$  (upper limit).

Statistical substitution of *J* boron atoms into an *n*-atom carbon cluster produces a distribution given by  $\rho(B_J C_{n-J}) / \rho(C_n) = \{n(n-1)\dots(n-J+1)\} / J! [B/C]^J$ . With the experimental B/C  $\sim 1/3$ , the statistical trimer distribution is

$$\rho(C_3) : \rho(BC_2) : \rho(B_2C) : \rho(B_3) \sim 1 : 1 : 0.33 : 0.03.$$

Agreement between distributions implies trimers form by random condensation of well-mixed atoms, uninfluenced by the relative energies of the trimers, the energies of their precursors, or preferential kinetics pathways that could otherwise distort the statistics.

Linear C<sub>3</sub> and cyclic BC<sub>2</sub>, disappeared entirely when the matrices were repeatedly annealed to temperatures between 25 K and 35 K, but cyclic B<sub>2</sub>C was inert.

Linear C<sub>4</sub> and BC<sub>3</sub> (BCCC) disappeared more slowly, and linear B<sub>2</sub>C<sub>2</sub> (BCCB) grew to  $\sim 95\%$  of its final value during the first annealing. Once formed, B<sub>2</sub>C<sub>2</sub>, like B<sub>2</sub>C, was also inert to further reaction.

The sources of  $B_2C_2$  are from condensation of atom plus trimer ( $B + BC_2$  but not  $C + B_2C$ ) or dimer + dimer ( $BC + BC$  but not  $B_2 + C_2$ ). Although  $BC$  was not observed, the upper limit of  $\rho(BC)$  is larger than  $\rho(B_2C_2)$  so that  $BC$  cannot be ruled out as a source of  $B_2C_2$ .

The growth of  $B_2C_2$  is conclusive evidence of the presence of  $BC$  and/or  $B$  in the originally deposited matrix in an amount at least as great as the growth of  $B_2C_2$ .

Linear  $C_5$ ,  $BC_4$  (BCCCC) and  $B_2C_3$  (BCCCCB) and larger linear clusters ( $B_JC_{n-J}$ ,  $5 < n < 11$ ,  $J = 0, 1, 2$ ), all grew upon annealing.

The sources of  $B_2C_3$  are dimer + trimer ( $BC + BC_2$  but not  $B_2 + C_3$ ) and atom + tetramer ( $B + BC_3$  but not  $C + B_2C_2$ ).

Since  $\rho(BC_2) \sim 5\rho(BC_3)$  in the initially deposited matrix, the  $BC + BC_2$  source is dominant. Growth of  $B_2C_3$  conclusively establishes the presence of  $BC$  in the matrix in an amount at least as great as the amount by which  $B_2C_3$  grows.

Growth of  $BC_4$  occurs primarily by  $BC + C_3$  rather than  $B + C_4$  or  $C + BC_3$  because  $\rho(C_3) \sim 10\rho(C_4)$  and  $\rho(C_3) \sim 2\rho(BC_3)$ . Growth of  $C_5$  occurs by  $C + C_4$  and  $C_2 + C_3$ , which establishes the presence of  $C$  and/or  $C_2$  in the original matrix in an amount at least as great as  $C_5$  growth.

Disappearance of triangular  $BC_2$  requires breaking of one of its B-C bonds when one of its carbon atoms is attacked. The major reorganization of electronic energy involved in opening the ring appears to occur with little ( $< \sim 3 \text{ kcal mol}^{-1}$ ) or no energy barrier, which makes this small molecule a candidate for an interesting *ab-initio* study of unusual reactivity at low temperature.

## Conclusions

Annealing kinetics of disappearance of  $C_3$  and  $BC_2$ , and of appearance of  $B_2C$ ,  $C_4$ ,  $BC_3$ ,  $B_2C_2$ ,  $C_5$ ,  $BC_4$ , and  $B_2C_3$  unequivocally establishes the presence of atoms and dimers in the originally deposited matrix.

$\sim 80\%$  or more of the initially deposited HEDM existed as atoms, dimers and trimers.

Molecules with two boron atoms are immune from radical attack and condensation during annealing.

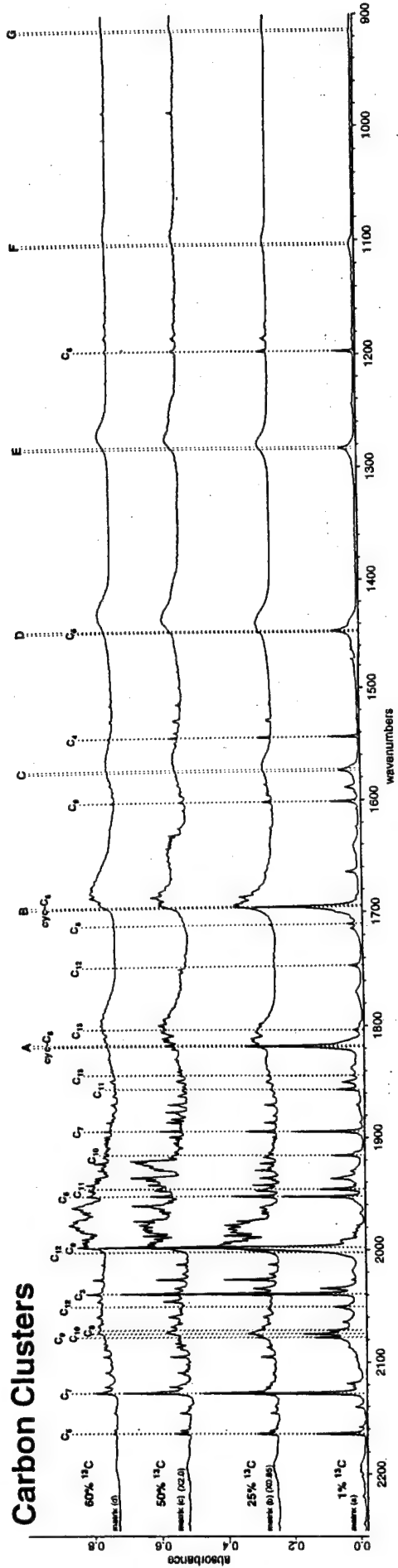
## Future Work

Continued development of source for production of higher flux beam of nearly pure boron atoms.

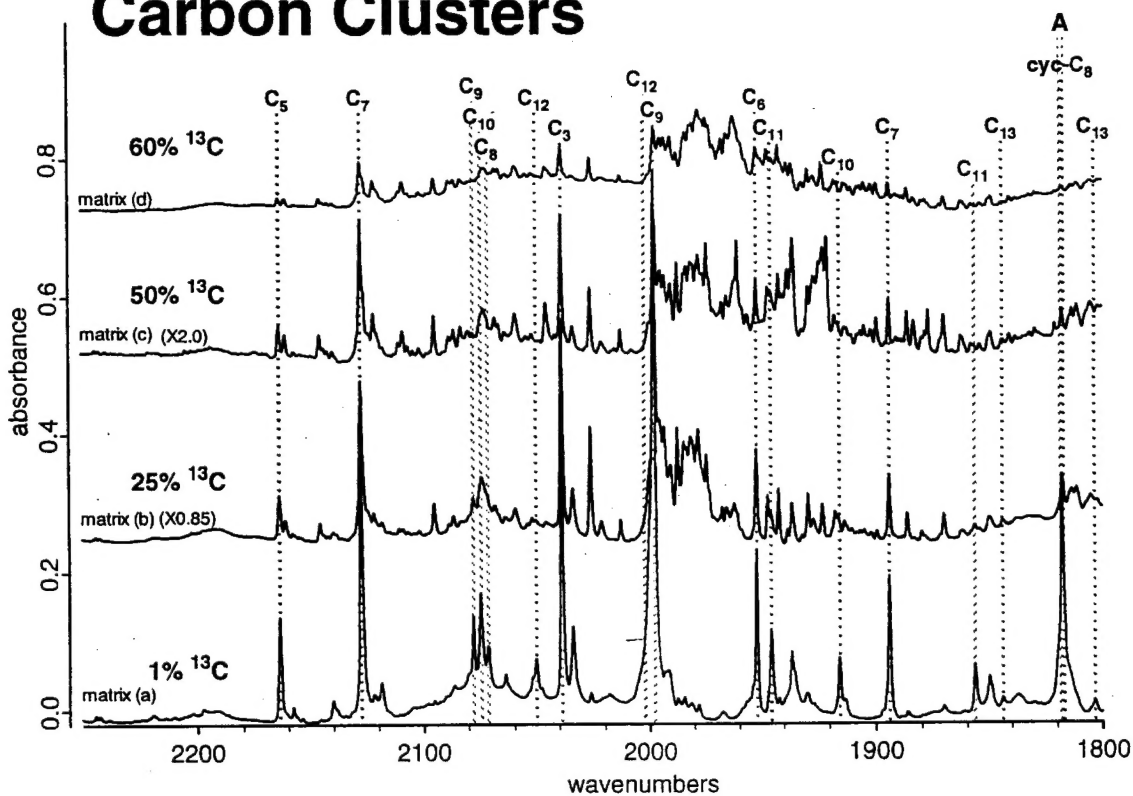
Map of "islands of stability" of pure boron HEDM;  $B_2$  or  $B_3$  may be the ultimate sink for atoms in the low temperature HEDM environment.

Determine reactivity of boron atoms with hydrogen during co-deposition.

Develop rapid condensation methodology to prevent reaction of  $B$  with  $H_2$ .

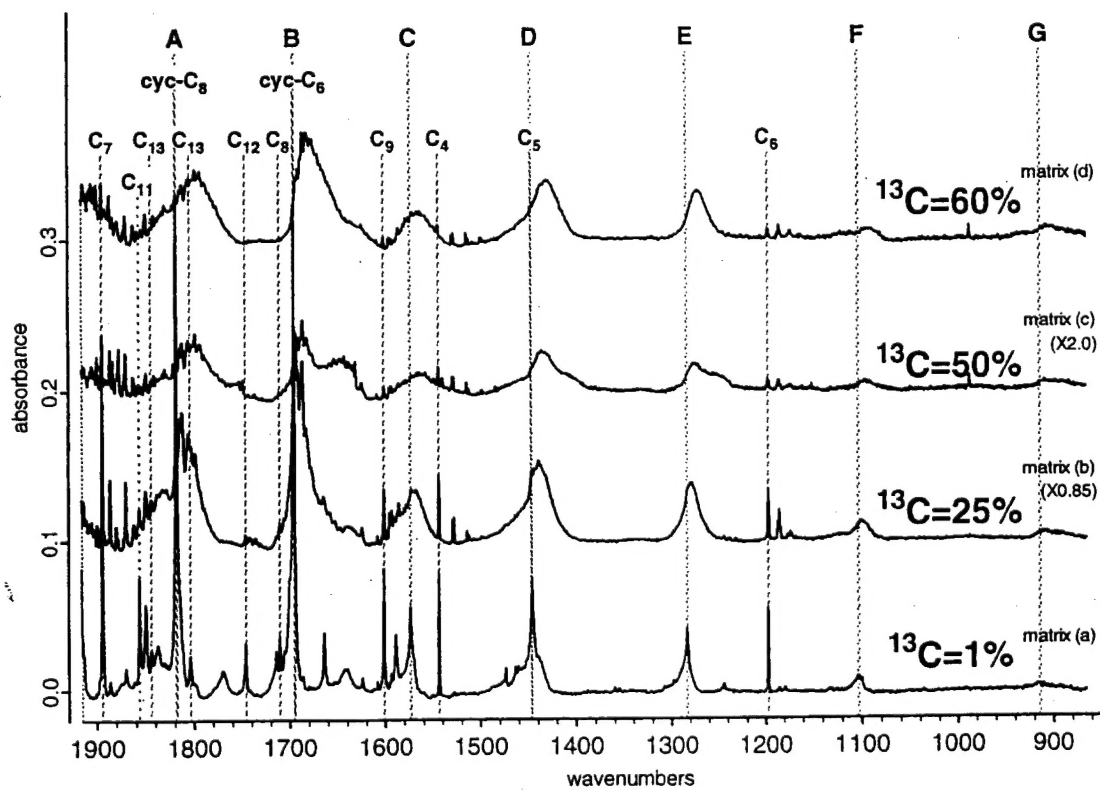


# Carbon Clusters



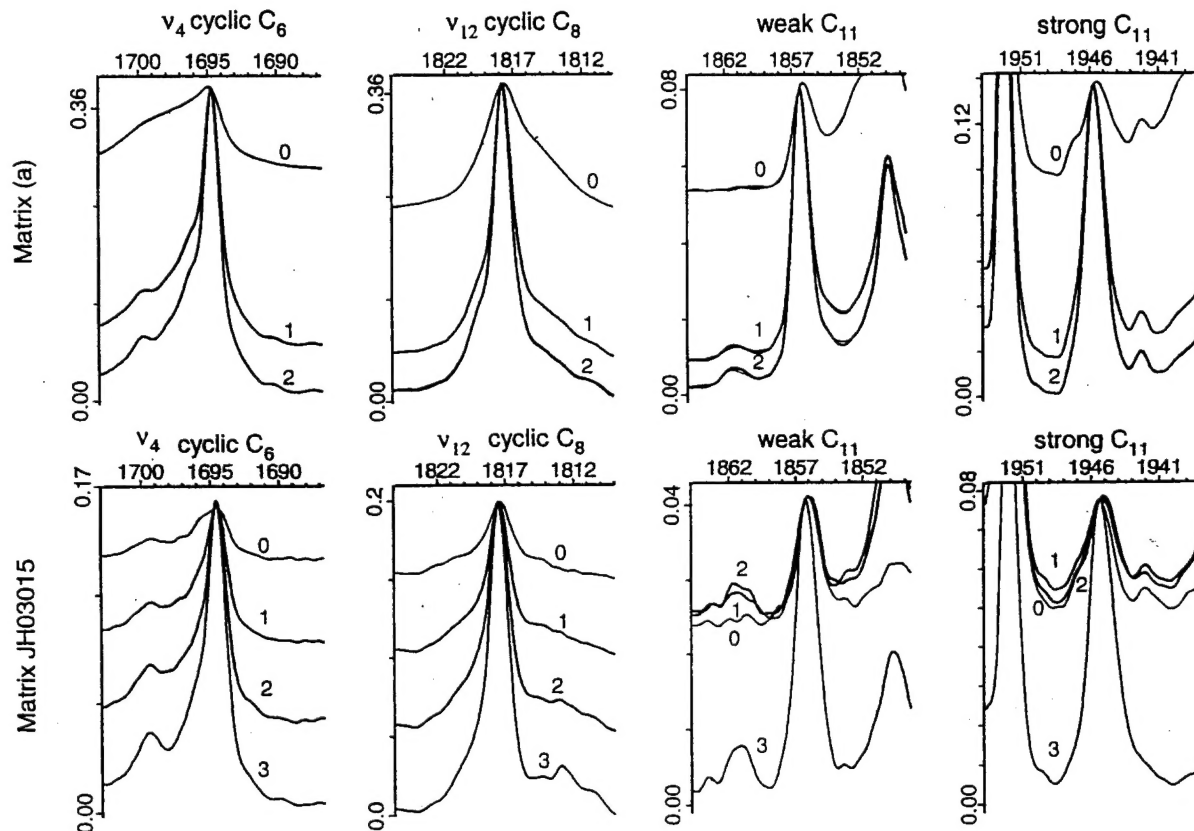
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# Carbon Clusters



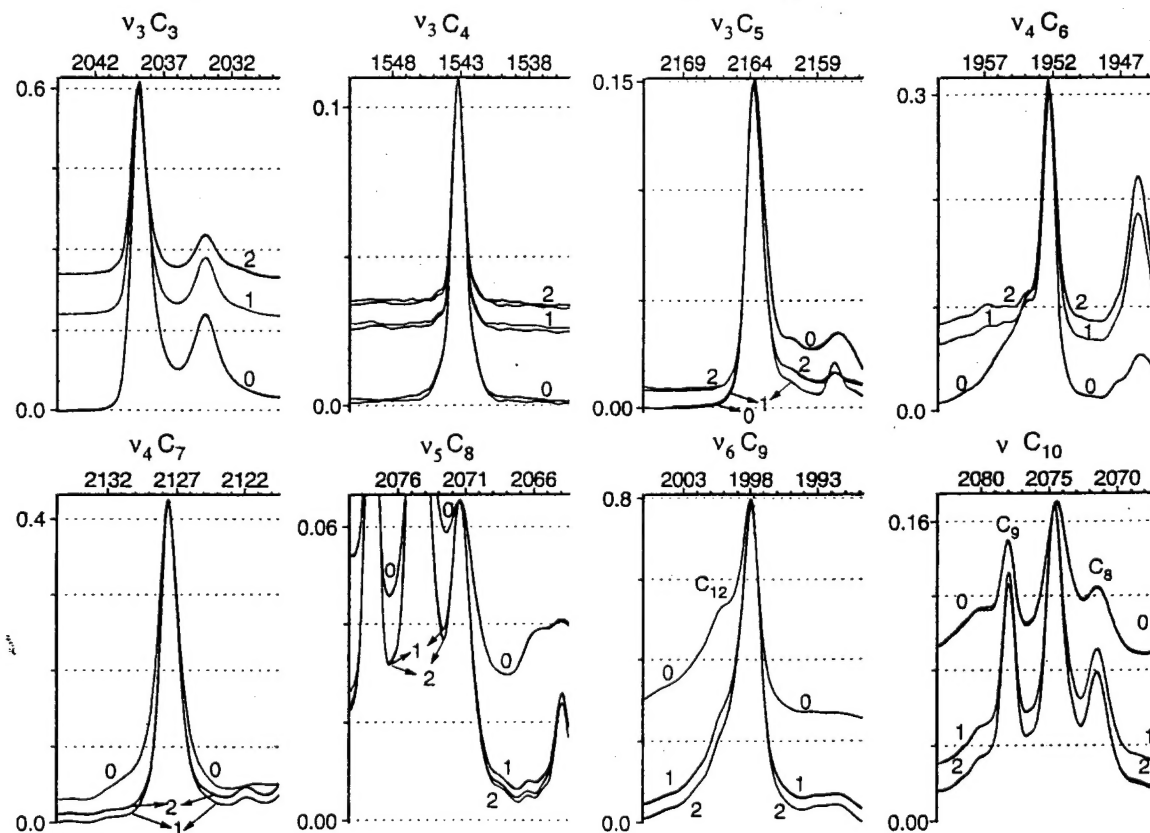
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## Growth upon annealing - $cC_6$ and $cC_8$ - two fundamentals of $C_{11}$



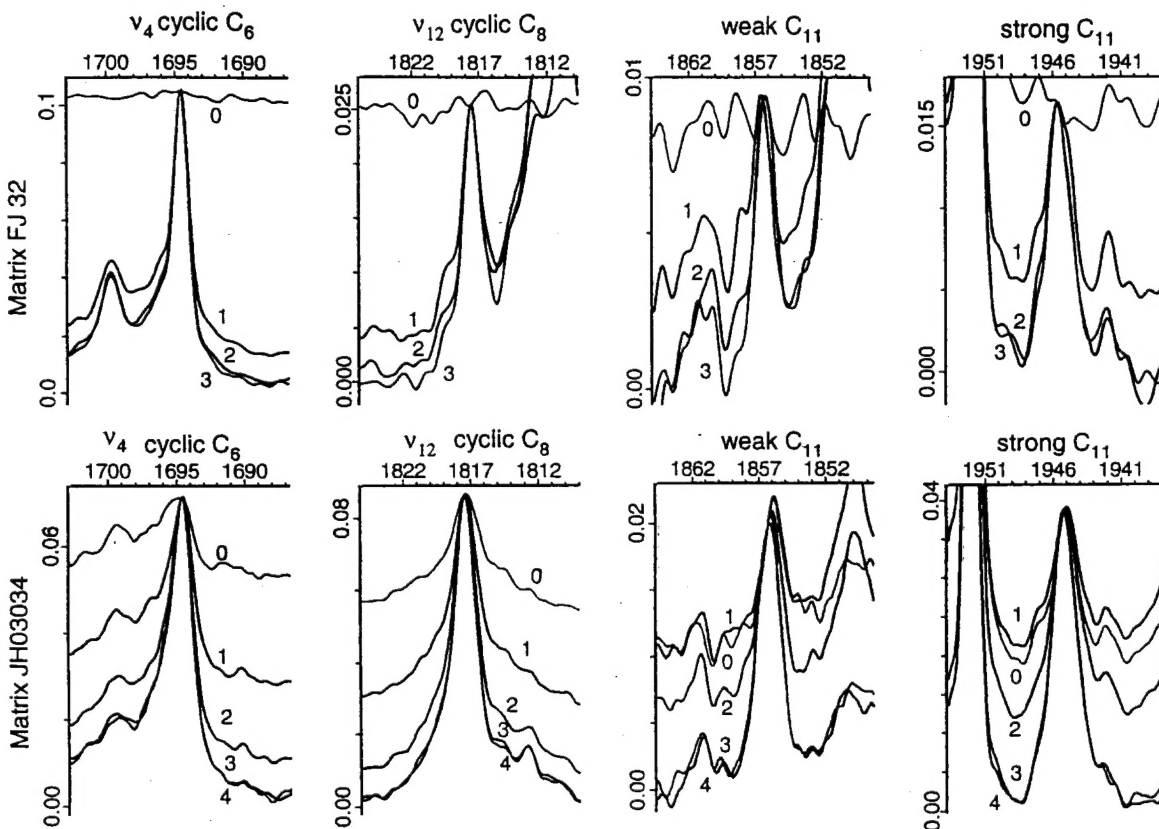
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## Carbon Matrix (a) - Linear $C_3$ to $C_{10}$ Clusters



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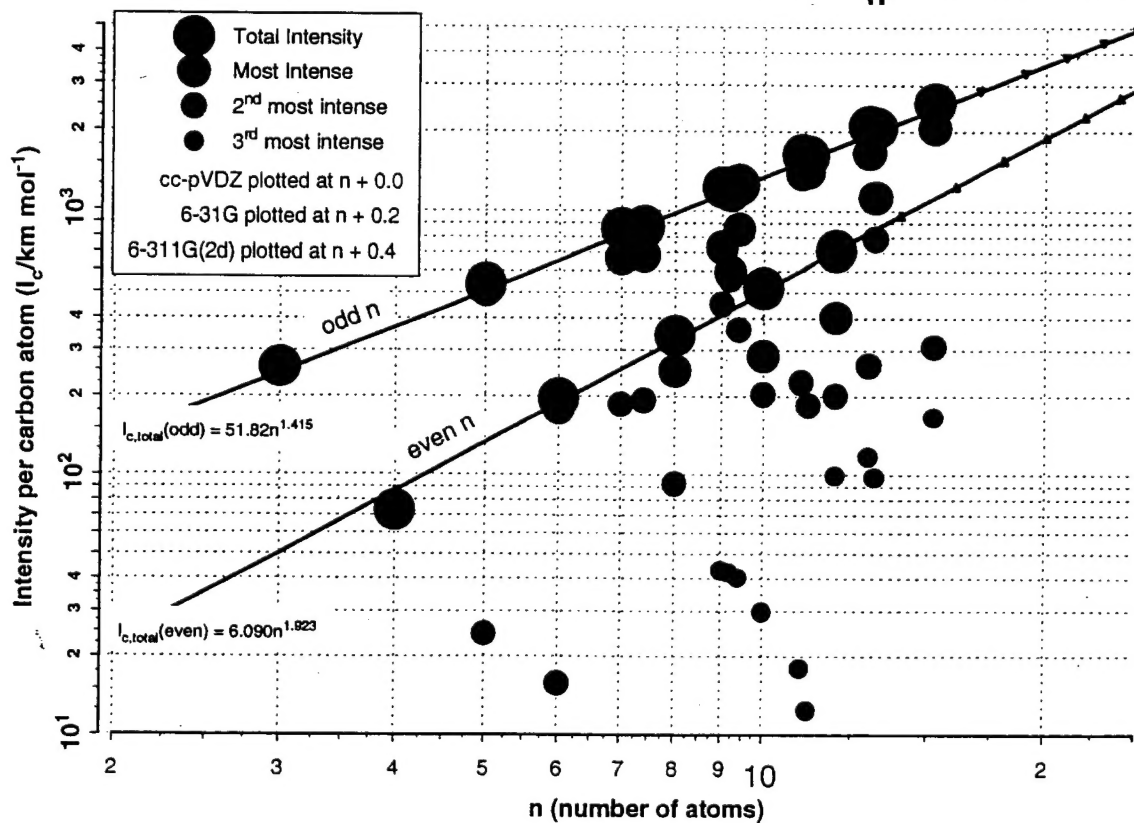




Carbon Matrix FJ 32, JA03034

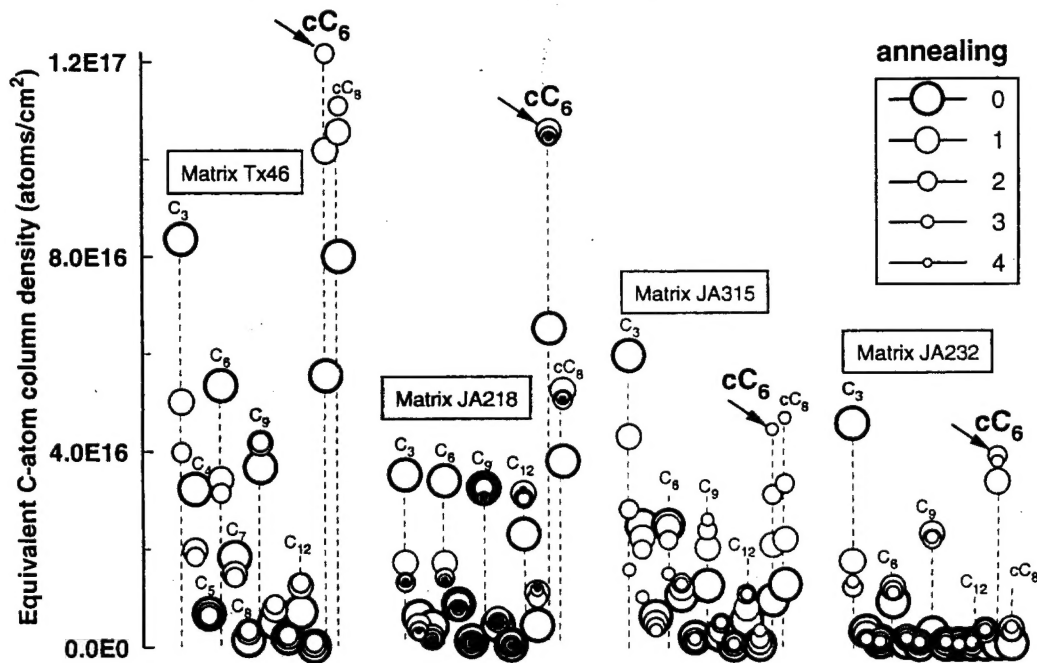
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## Theoretical Infrared Intensities of Linear $C_n$ , DFT/B3LYP



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# Carbon cluster distributions



Most of the carbon condenses to cC<sub>6</sub> and cC<sub>8</sub> in all matrices

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## Conclusions from Carbon HEDM Research

**Quantitative analysis** - Establishes HEDM density, distribution of carbon clusters, tracking of growth and decay of carbon clusters, carbon bookkeeping - quantification of "invisible carbon", C-atom and C<sub>2</sub>.

Highest density matrix (equivalent C-atom density ~ 1 mole percent in argon) contained 40% "invisible" carbon (C, C<sub>2</sub>), determined by tracking the growth of the "visible" (measurable) carbon to a constant composition after repeated annealing. Main product of condensation is cyclic C<sub>6</sub>.

Yields of cyclic-C<sub>6</sub> are a factor of two larger than the combined yield of all other clusters in the fully condensed, highest density matrices. Cyclic-C<sub>6</sub> is the dominant condensation product.

Substrate must be shielded from oven to prevent condensation during deposition. Higher temperature oven places higher heat load on substrate, which promotes condensation.

Obtained higher density matrices by decreasing argon flux and maintaining oven flux. However, condensation was also increased.

Matrices produced with argon/5% H<sub>2</sub> caused nearly complete loss of C<sub>n+1</sub> and C<sub>n+2</sub> relative to C<sub>n+3</sub>, suggesting that H<sub>2</sub> scavenges C-atoms efficiently during co-deposition.